

# Organocatalyzed Enantioselective Synthesis of Chiral Heterocyclic Compounds using Domino Cycloaddition/Rearrangement

## CiHS Oral Presentation



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# Introduction

## Asymmetric synthesis?

- ✓ Asymmetric synthesis is synthesis of optically active compounds, preferably single enantiomers.

### Organocatalyst

- ✓ No air & moisture sensitive
- ✓ Inexpensive
- ✓ Nontoxic
- ✓ Easy screening

### Transition Metal catalyst

- ✓ Tedious process
- ✓ Air and moisture sensitive.
- ✓ Metal impurity's in final product

### Biocatalysts

- ✓ Limited choice of substrate
- ✓ Formation of single enantiomer.

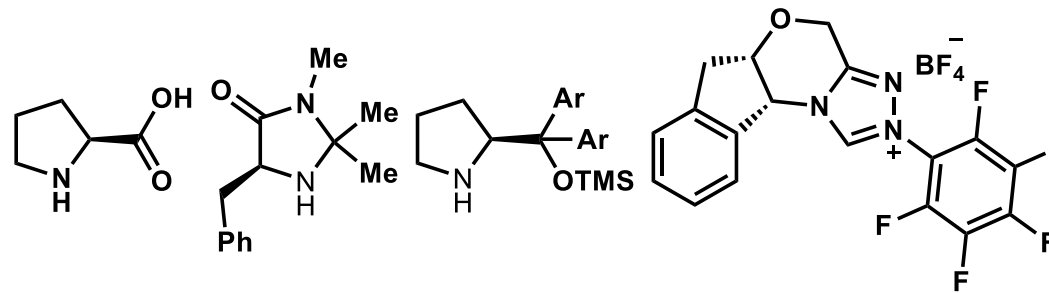
## Organocatalyst?

- ✓ The catalysis with small organic molecules, where an metals is not part of the active principle.

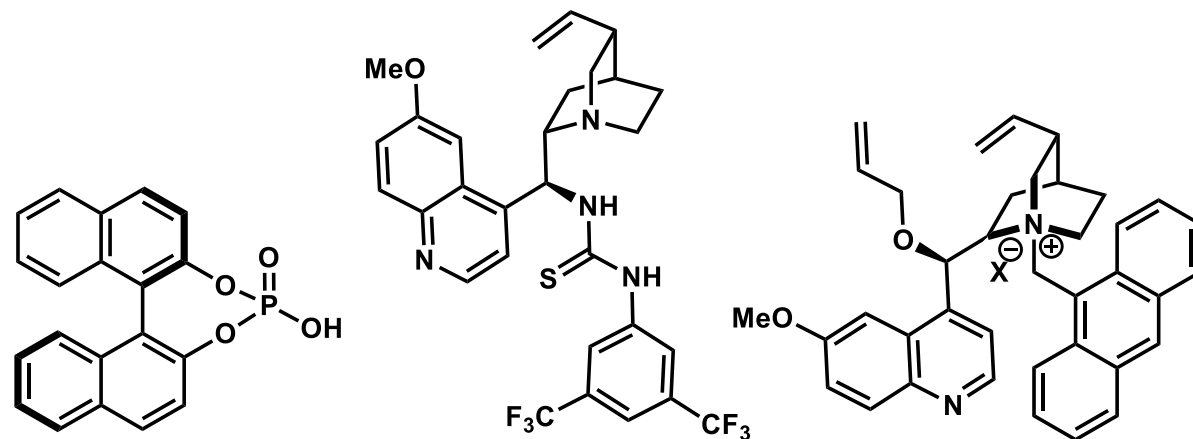
Lewis acid  
Lewis base

Branstad acid  
Branstad base

## Covalent bond activation

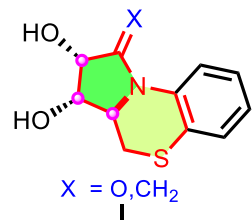
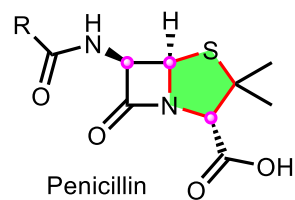


## Non-covalent bond activation

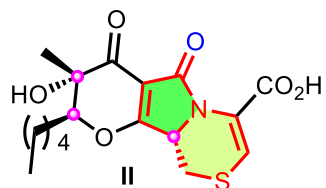


# Literature reports

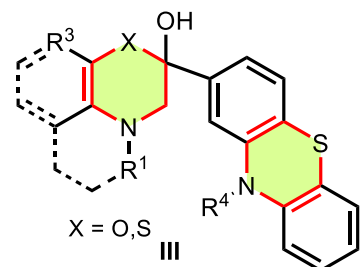
## Importance of thiazoles & pyrrolo-thiazines



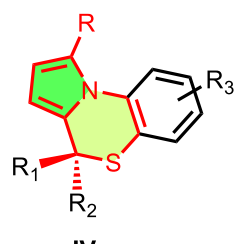
Inhibition of  $\beta$ -Ribosidases



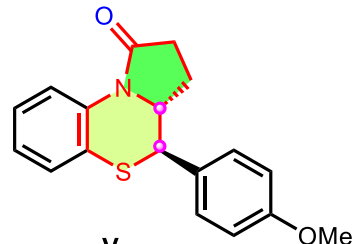
Naturel STAT<sub>3</sub> inhibitor



Antioxidant, anti-inflammatory, (SQS) inhibitory/hypolipidemic activities



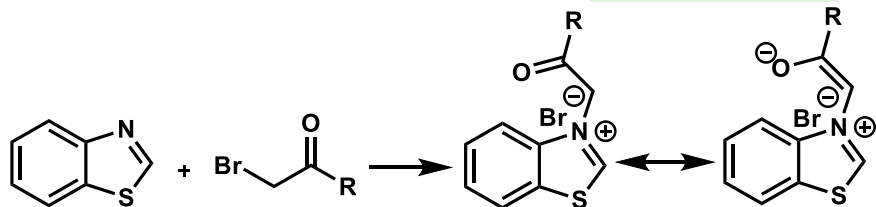
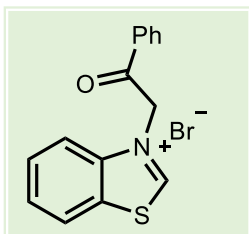
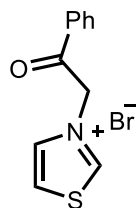
Ca entry blockers inhibitors



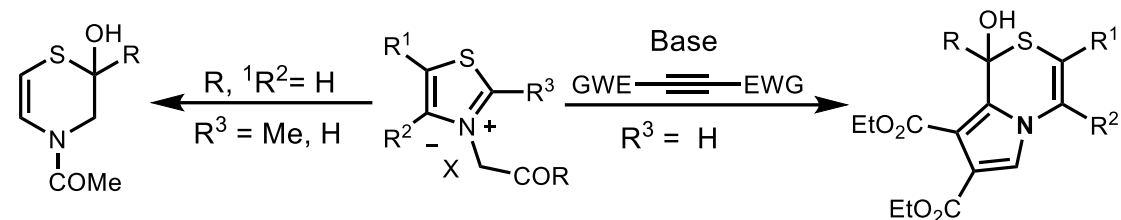
selective myocardial calcium channel modulators

A. Daich. *et al*, *Chem. Eur. J.* **2019**, *25*, 6113–6118

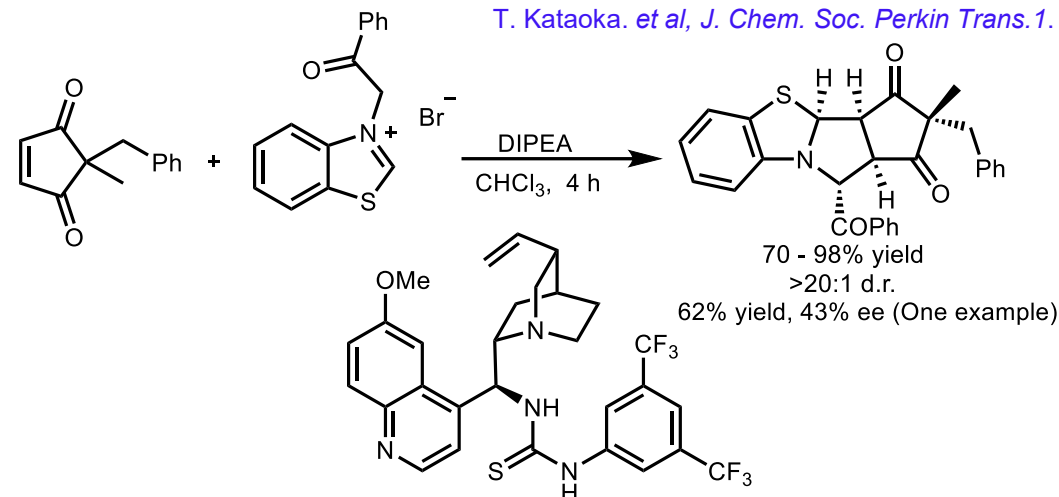
## Azomethine ylide



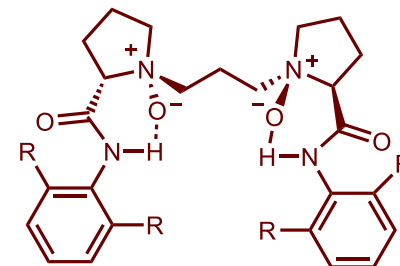
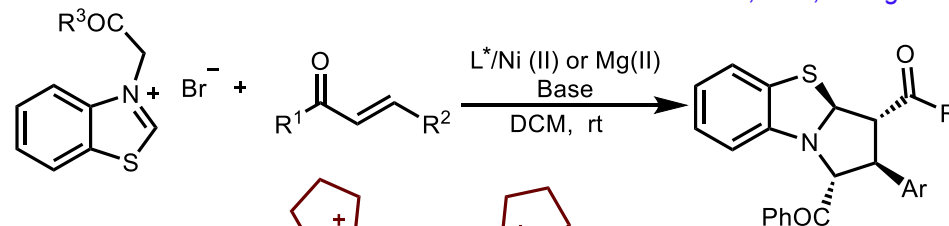
A. Naeem. *et al*, *Molecules.* **2016**, *21*, 1054



T. Kataoka. *et al*, *J. Chem. Soc. Perkin Trans. 1.* **1996**, 629 -635



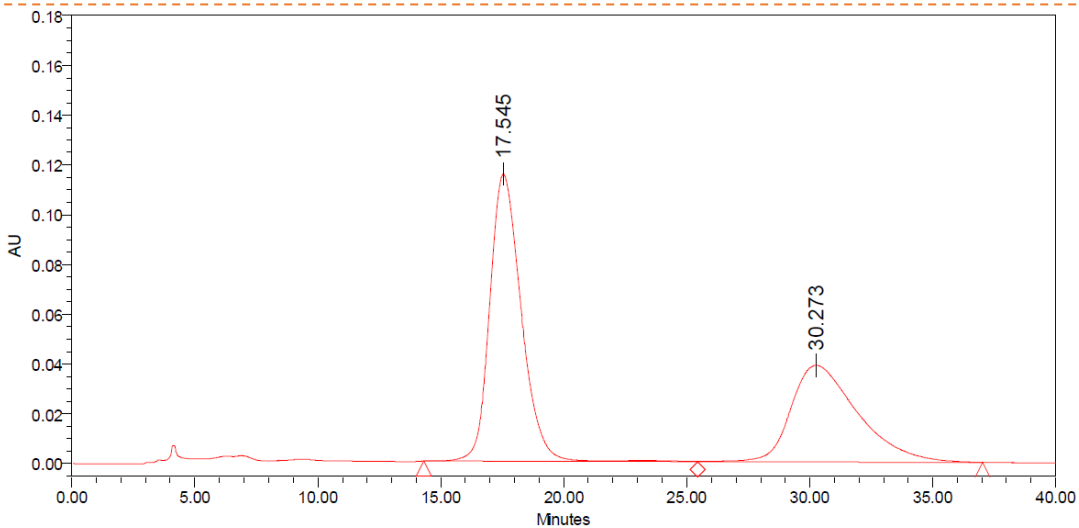
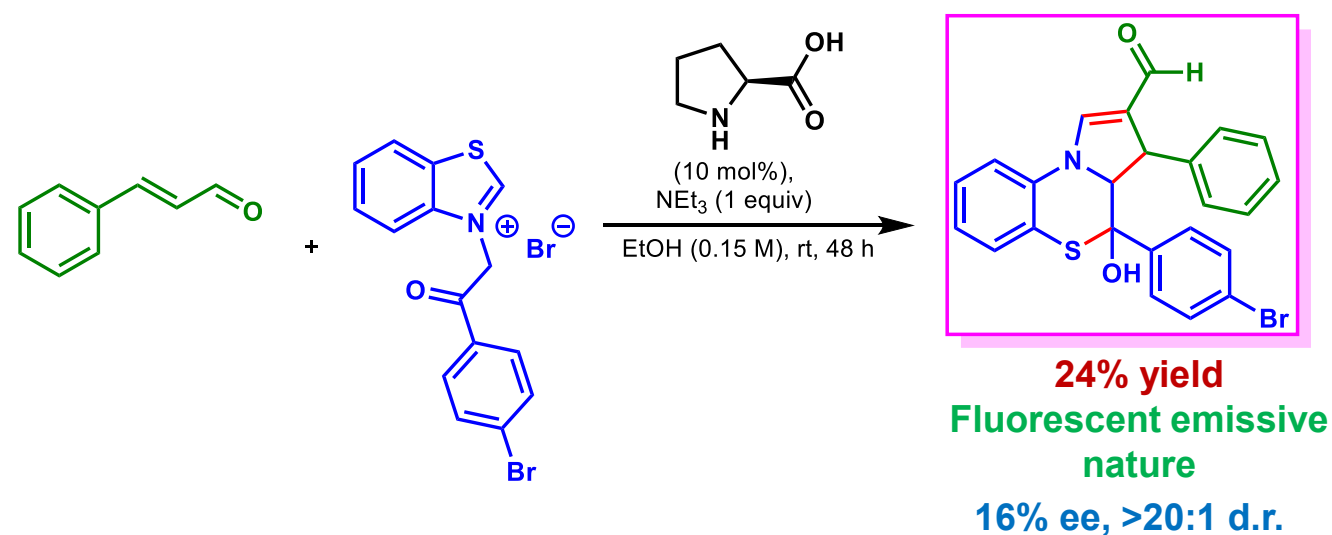
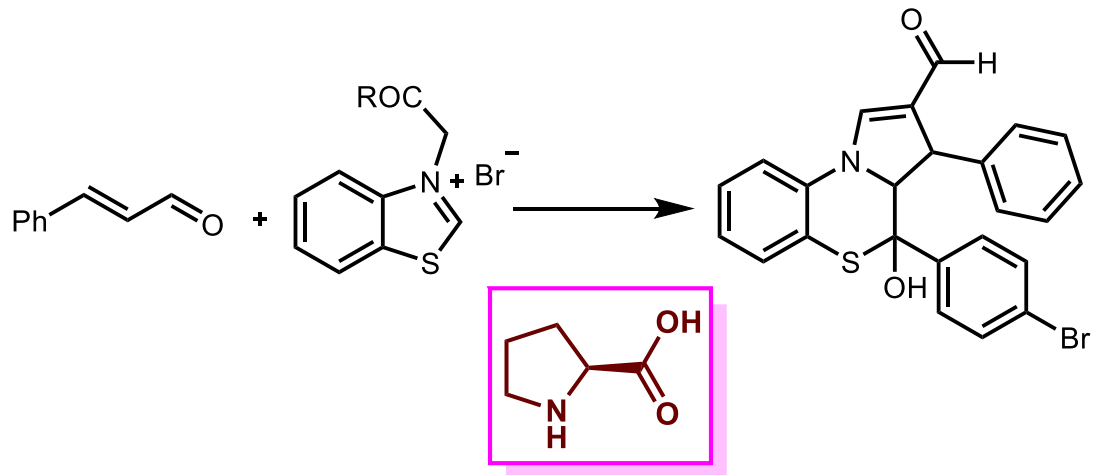
S. C. Pan, *et al*, *J. Org. Chem.* **2017**, *82*, 12763 - 12770



Chiral N,N'-dioxide ligand

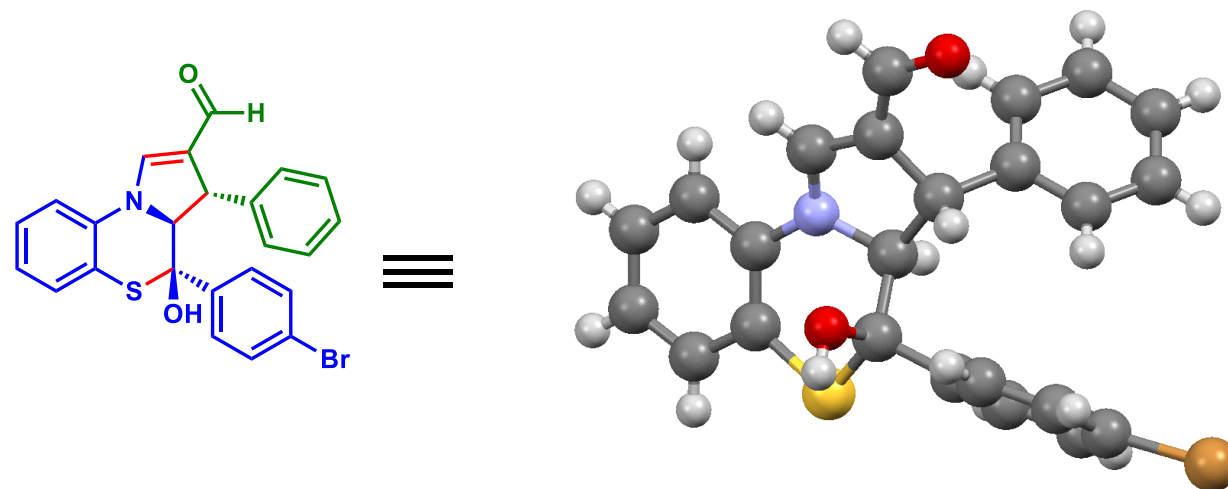
X. Feng, *et al*, *Org. Chem. Front.* **2018**, *5*, 2126 – 1231

# Our methodology

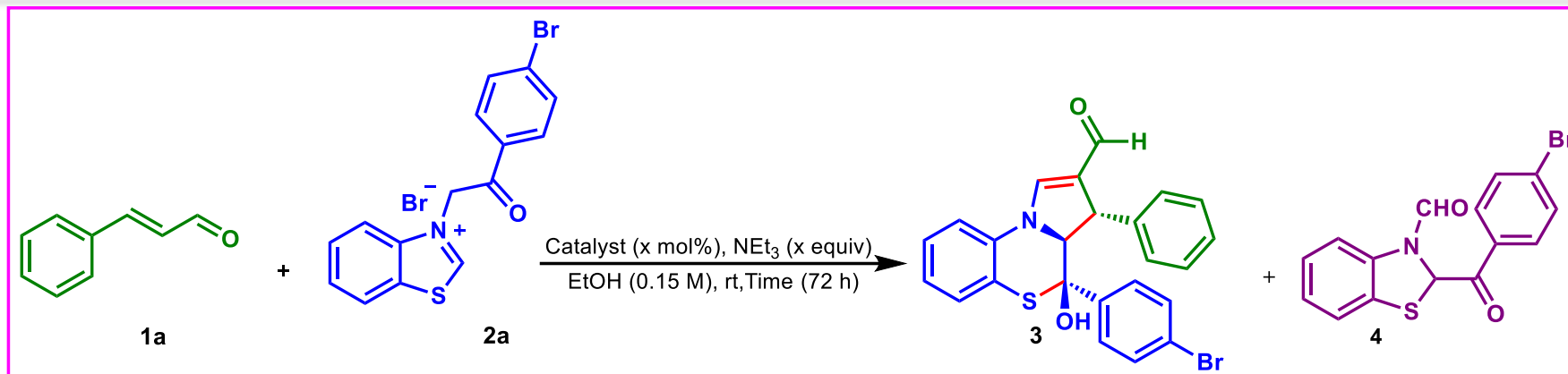


Peak Results

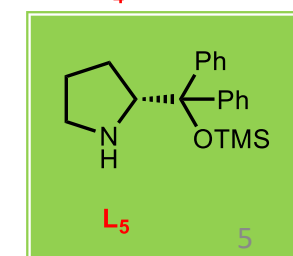
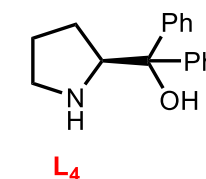
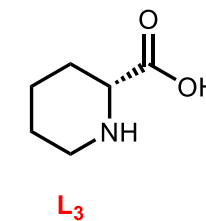
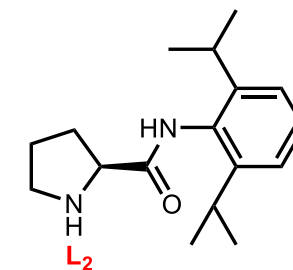
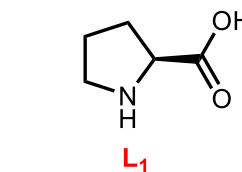
Name	RT	Area	Height	% Area	% Height
1	17.545	10289469	115420	58.4924	74.8420
2	30.273	7301666	38798	41.5076	25.1580
Sum		17591134.8		100.0000	100.0000



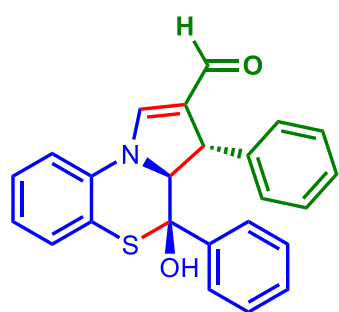
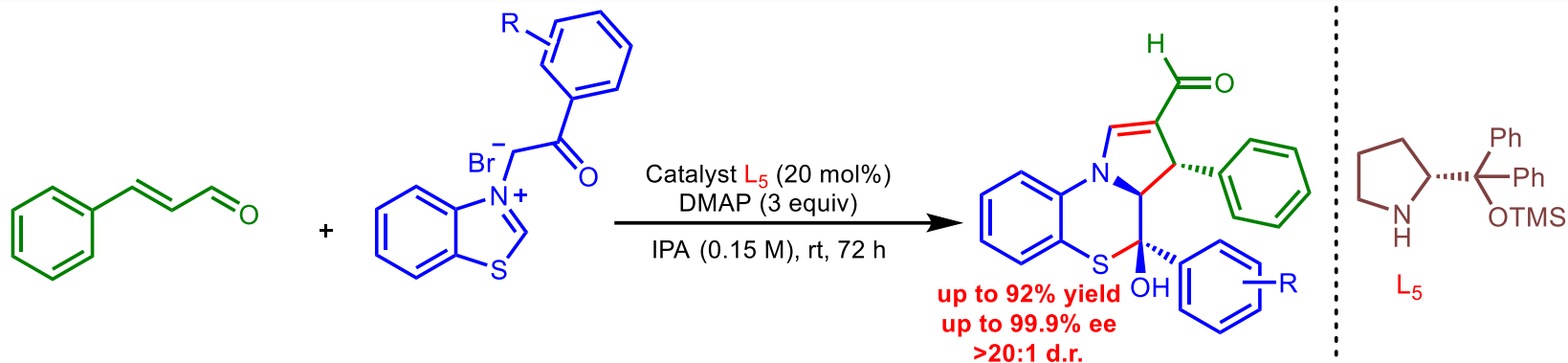
# Reaction optimization



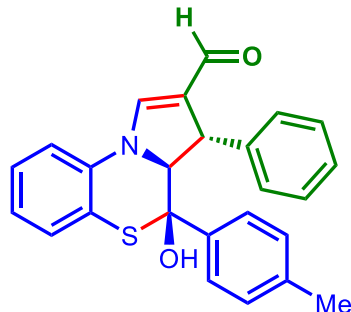
S. No	Base (equiv)	Catalyst (mol%)	Solvents (0.15M)	Yield (%)	ee (%)	d.r.	4 (%)
1	NEt <sub>3</sub> (1)	L <sub>1</sub> (20)	EtOH	24	24	19.1	10
2	NEt <sub>3</sub> (2)	L <sub>1</sub> (20)	EtOH	60	16	19.1	5
3	NEt <sub>3</sub> (3)	L <sub>1</sub> (20)	EtOH	75	16	19.1	trace
4	NEt <sub>3</sub> (3)	L <sub>4</sub> (20)	EtOH	32	38	19.1	10
5	NEt <sub>3</sub> (3)	L <sub>5</sub> (20)	EtOH	52	90	19.1	5
6	Dibutyl amine (3)	L <sub>5</sub> (20)	EtOH	95	10	19.1	-
7	DMAP (3)	L <sub>5</sub> (20)	EtOH	72	98	19.1	5
8	DMAP (3)	L <sub>5</sub> (20)	MeOH	48	74	19.1	5
9	DMAP (3)	L <sub>5</sub> (20)	DCM	65	98	19.1	10
10	DMAP (3)	L <sub>5</sub> (20)	IPA	87	99	19.1	trace



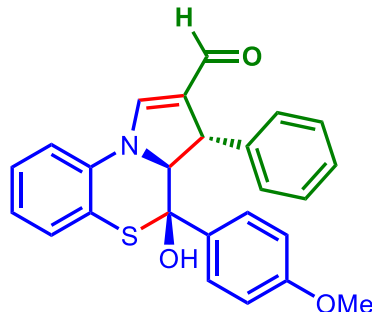
# Scopes of benzothiazolium salts



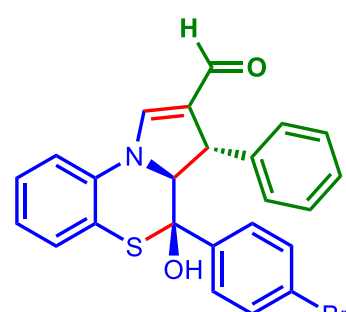
**3a**; 75%, 98% ee  
>20:1 d.r.



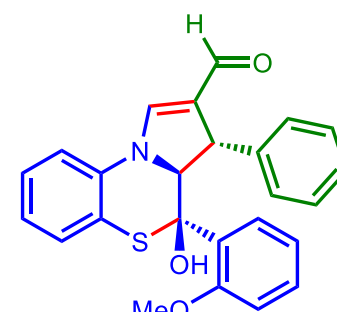
**3b**; 60%, 96% ee  
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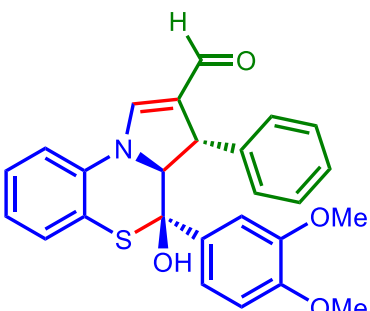
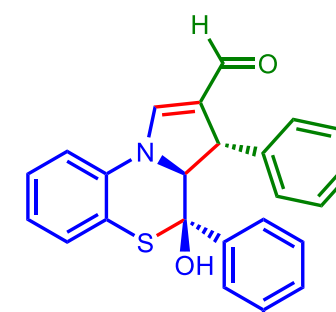
**3c**; 72%, 98% ee  
>19:1 d.r.



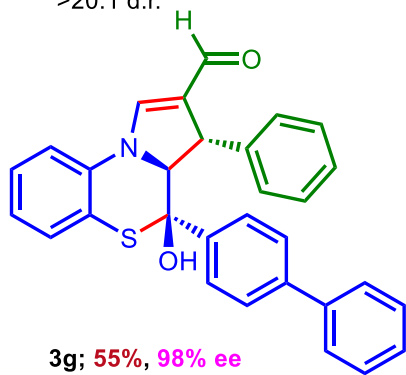
**3d**; 86%, 98% ee  
>19:1 d.r.



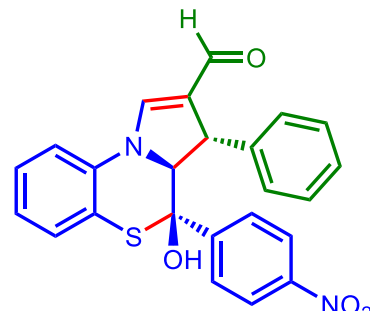
**3e**; 80%, 98% ee  
>20:1 d.r.



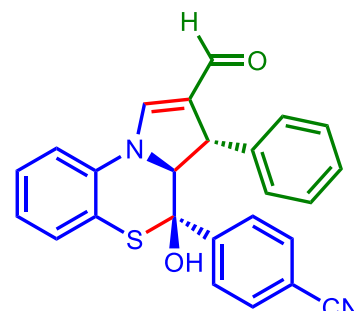
**3f**; 60%, 99% ee  
>20:1 d.r.



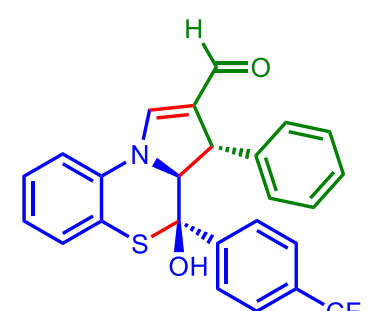
**3g**; 55%, 98% ee  
>20:1 d.r.



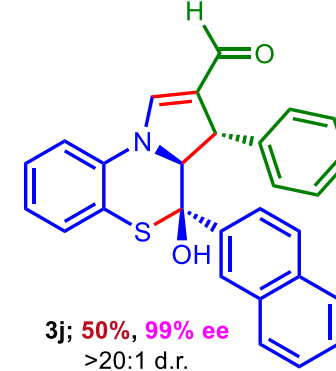
**3h**; 78%, 96% ee  
>20:1 d.r.



**3i**; 92%, 98% ee  
>20:1 d.r.



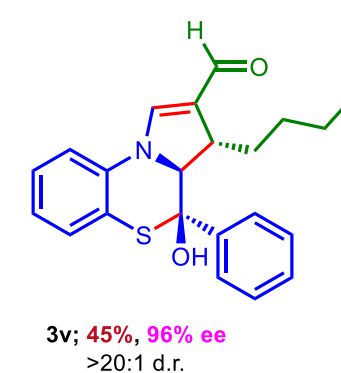
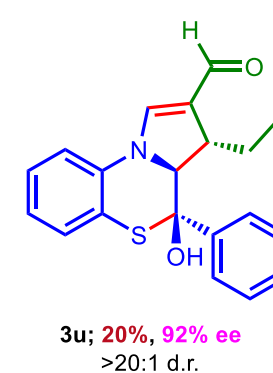
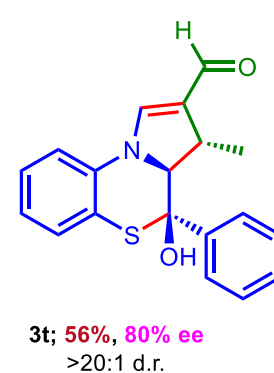
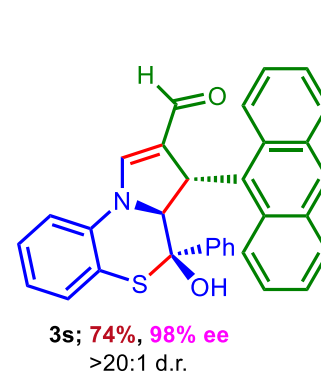
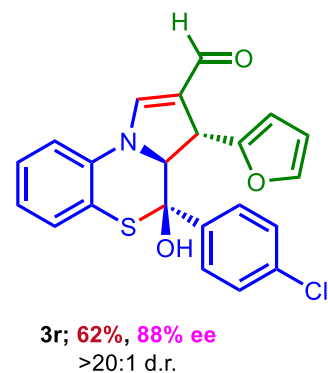
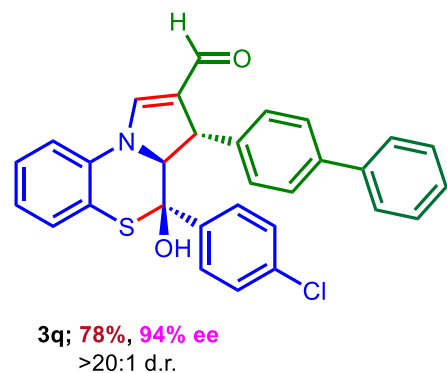
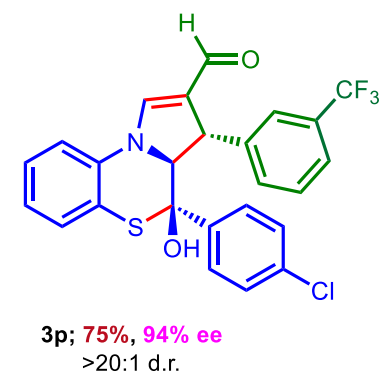
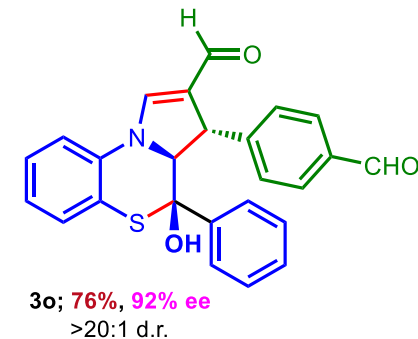
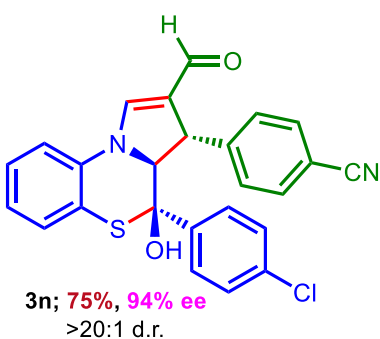
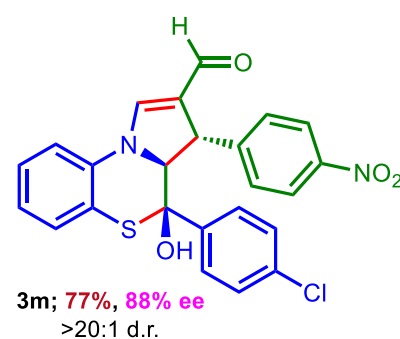
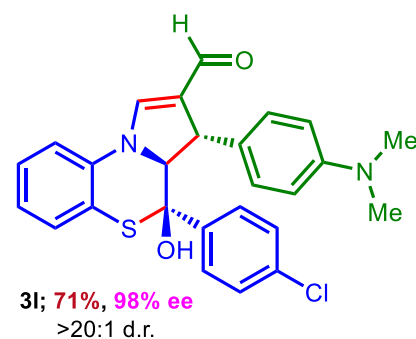
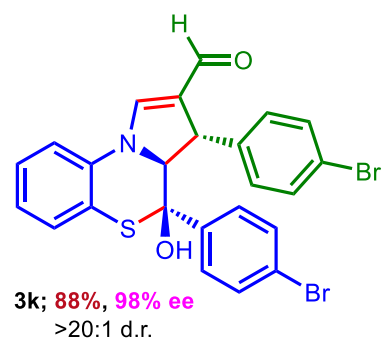
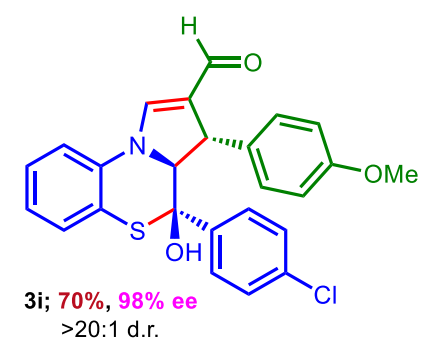
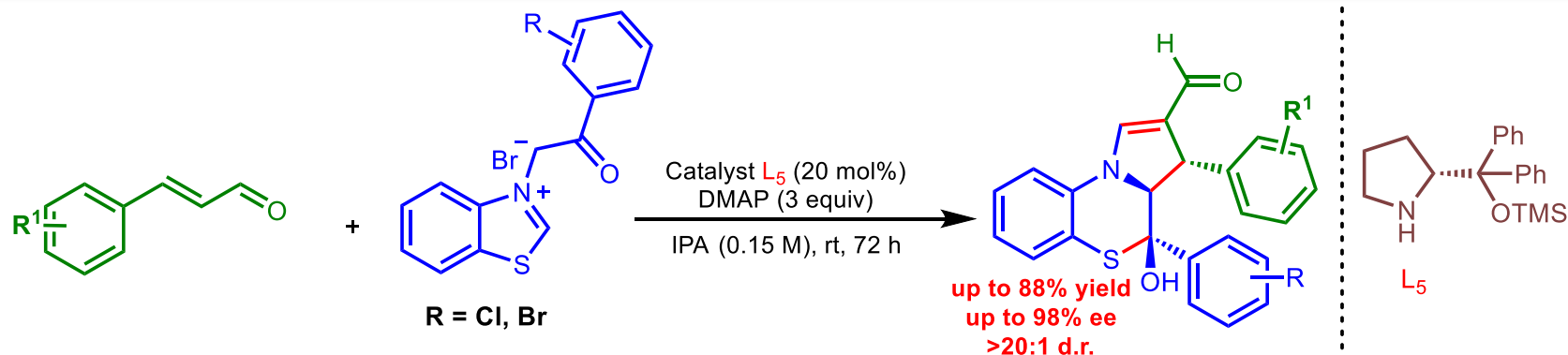
**3j**; 80%, 99% ee  
>20:1 d.r.

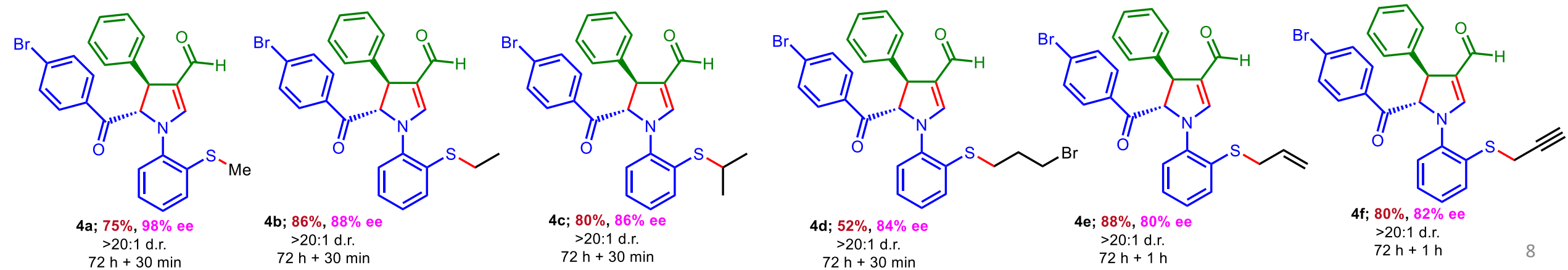
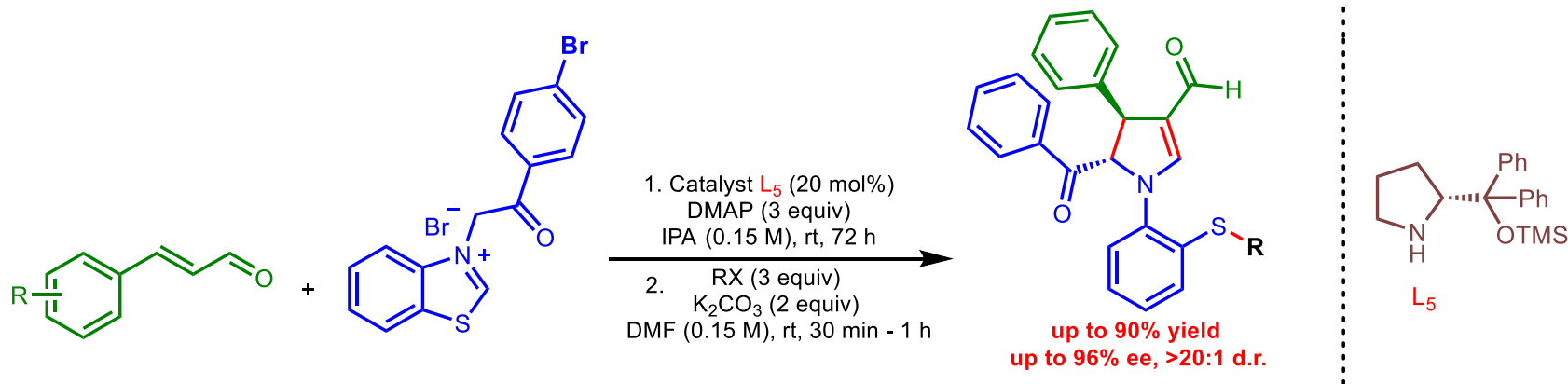
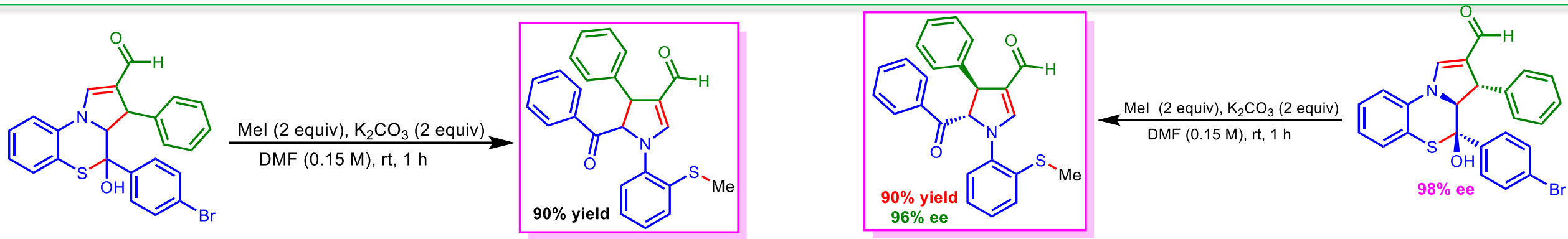


**3j**; 50%, 99% ee  
>20:1 d.r.

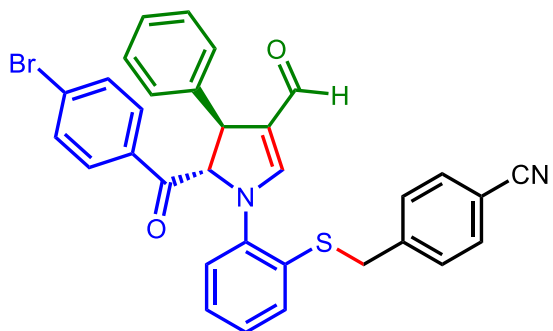
“Manuscript under revision”

# Scopes of $\alpha,\beta$ -unsaturated aldehydes

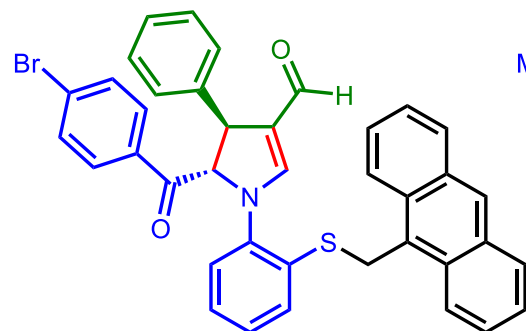




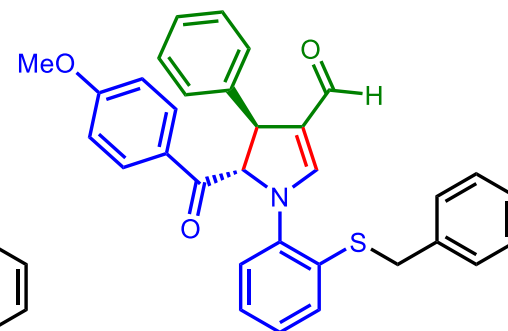
# Scopes of pyrrole-3-carbaldehyde



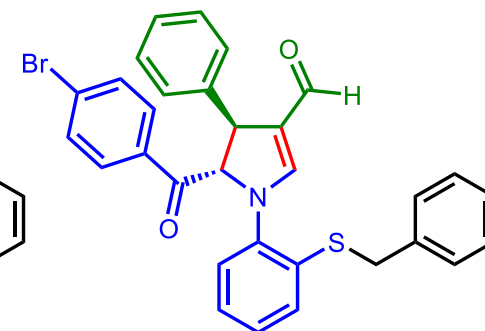
**4l; 70%, 96% ee**  
>20:1 d.r.  
72 h + 1 h



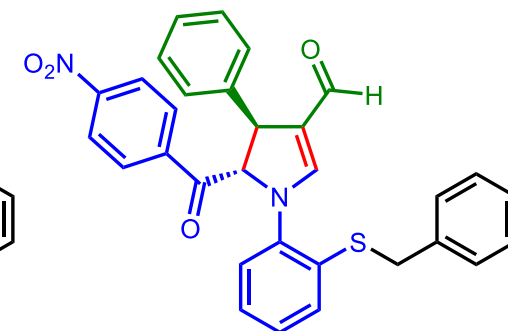
**4n; 85%, 90% ee**  
>20:1 d.r.  
72 h + 1 h



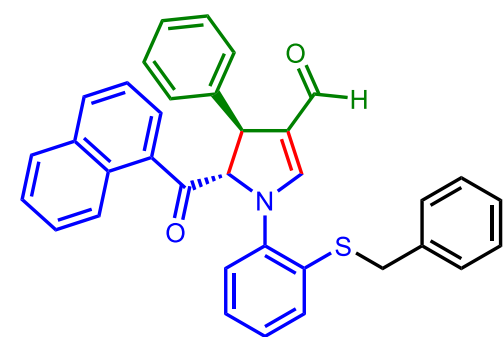
**4p; 72%, 96% ee**  
>20:1 d.r.  
72 h + 1 h



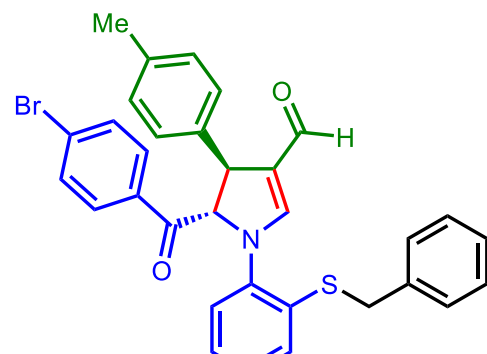
**4q; 85%, 98% ee**  
>20:1 d.r.  
72 h + 40 min



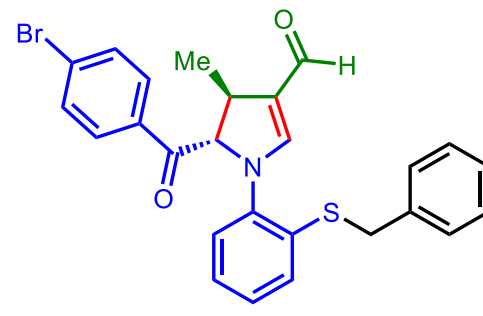
**4r; 75%, 90% ee**  
>20:1 d.r.  
72 h + 1 h



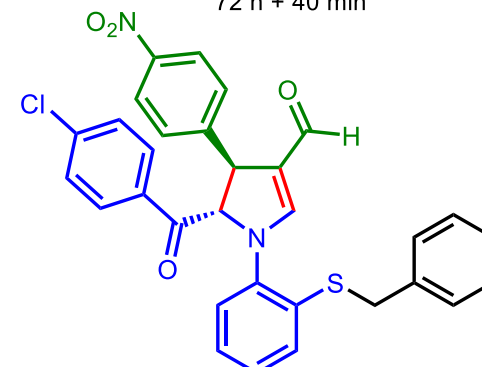
**4s; 60%, 90% ee**  
>20:1 d.r.  
72 h + 1 h



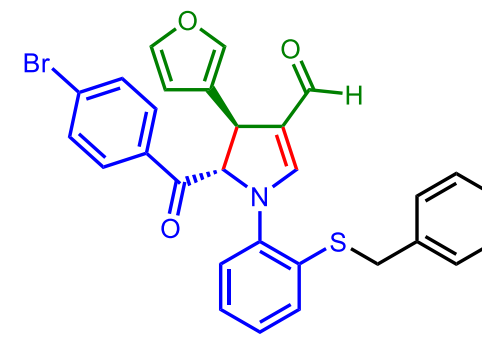
**4t; 76%, 98% ee**  
>20:1 d.r.  
72 h + 1 h



**4u; 48%, 82% ee**  
>20:1 d.r.  
72 h + 30 min

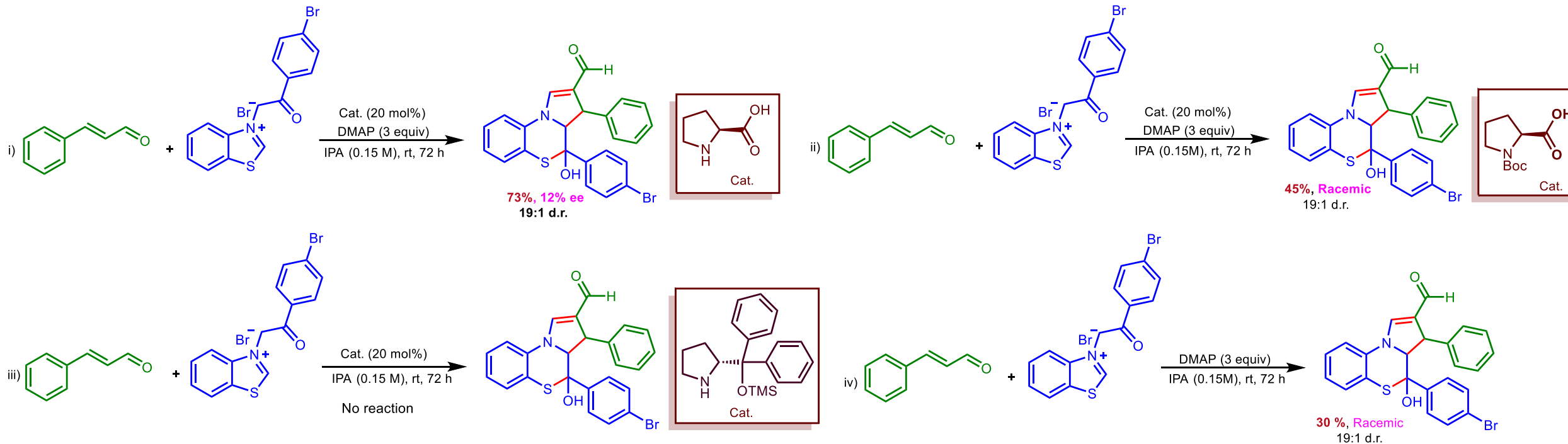


**4w; 72%, 98% ee**  
>20:1 d.r.  
72 h + 1 h

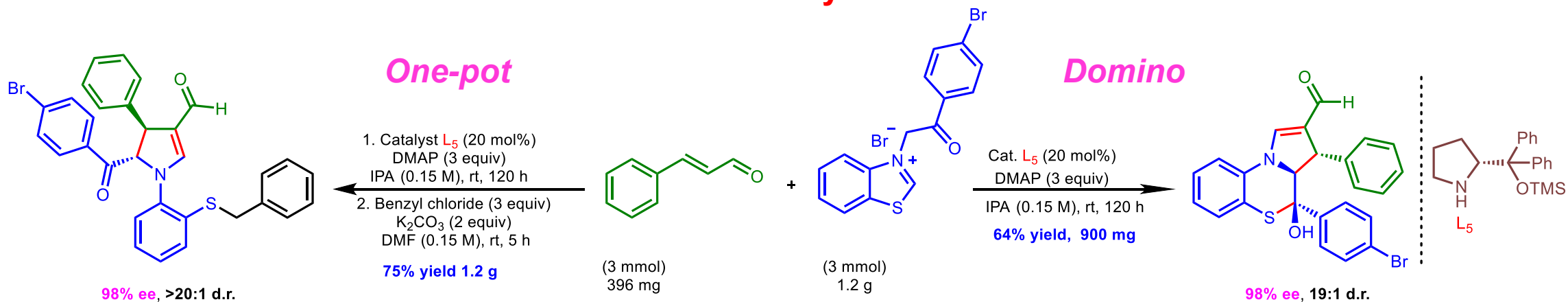


**4x; 65%, 88% ee**  
>20:1 d.r.  
72 h + 1 h

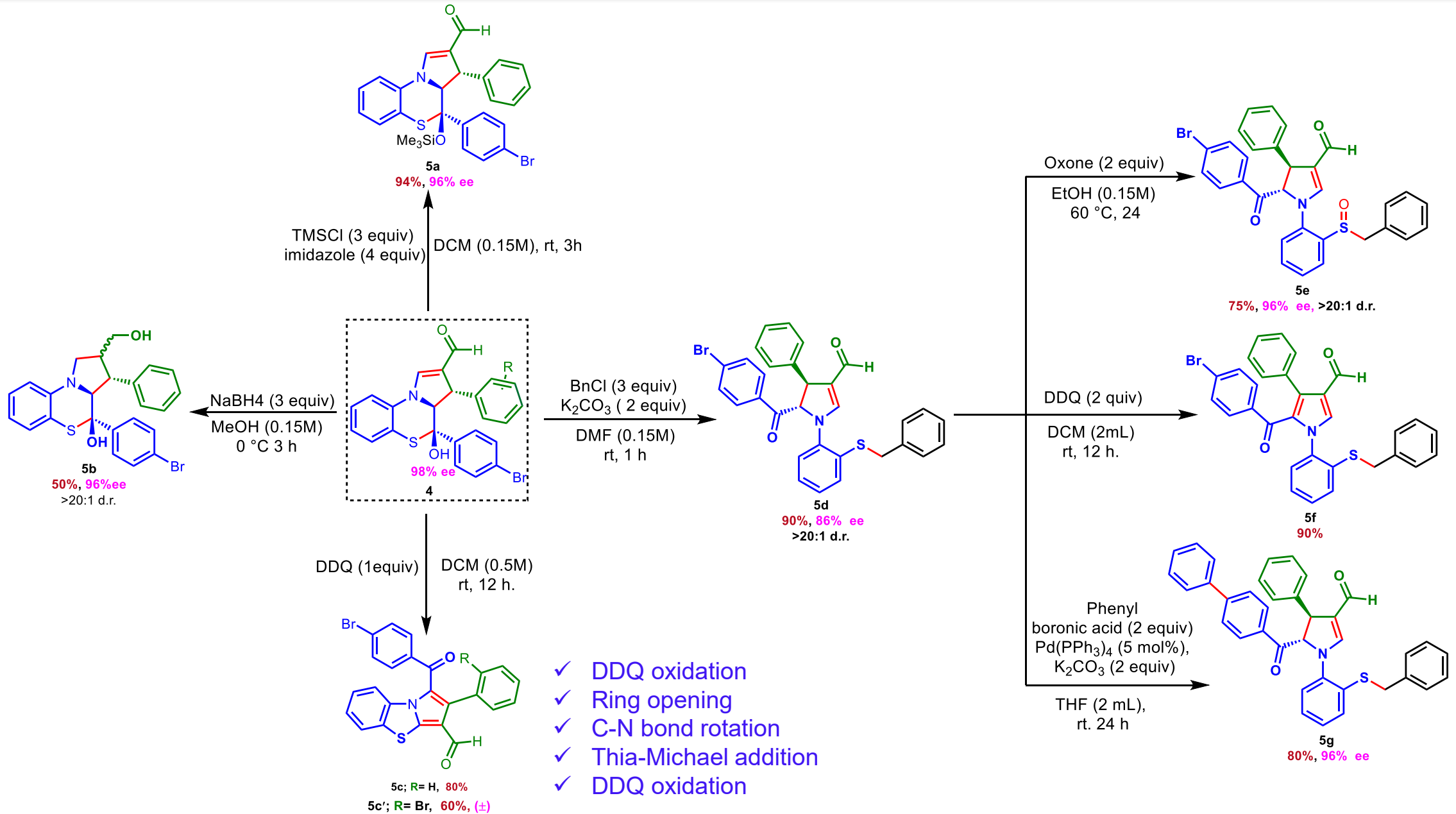
# Control experiments



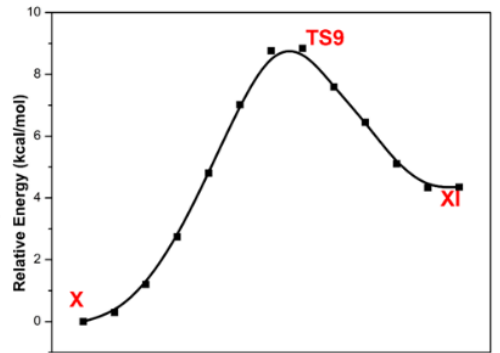
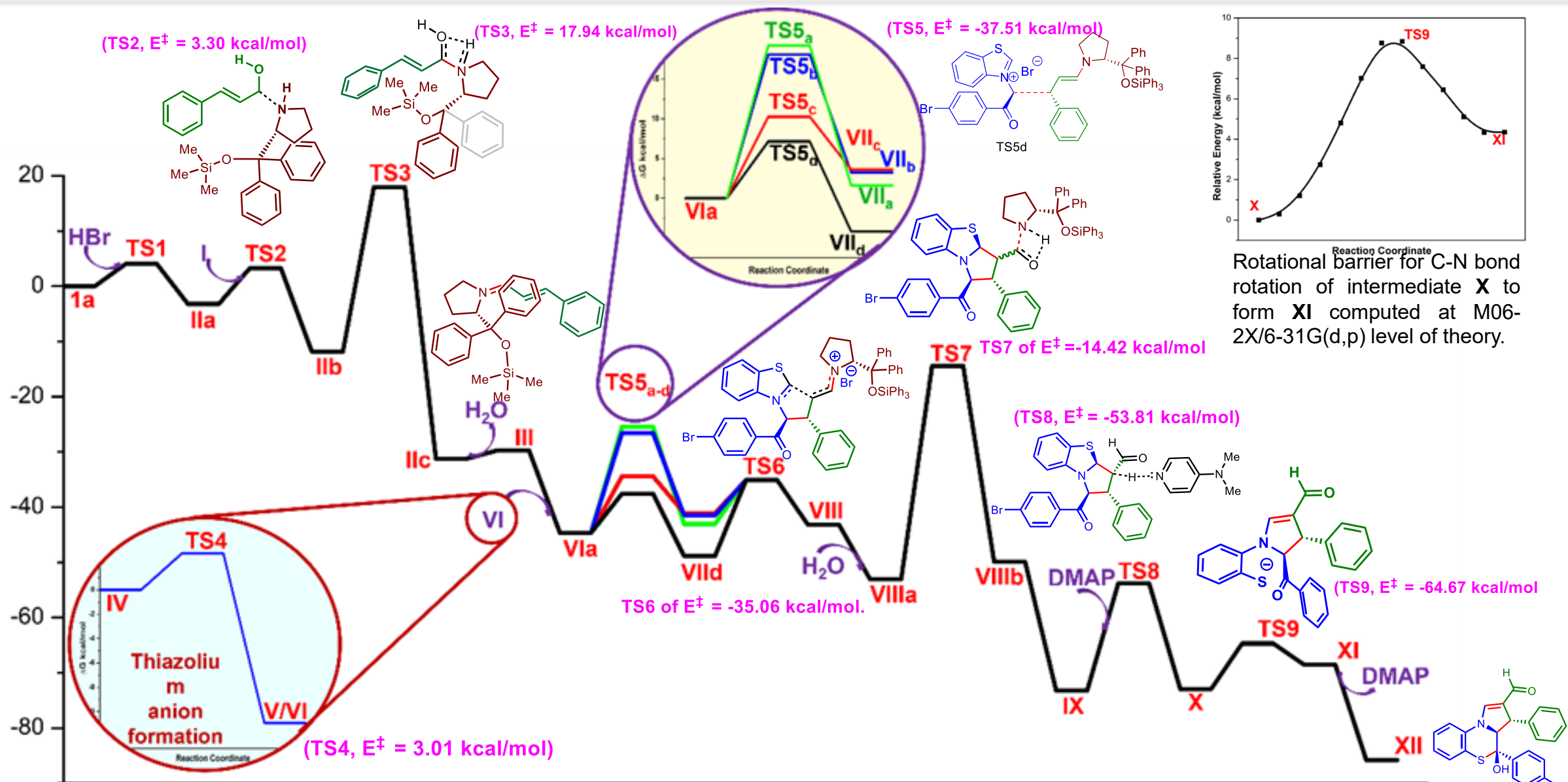
## Gram scale synthesis



# Synthetic utility



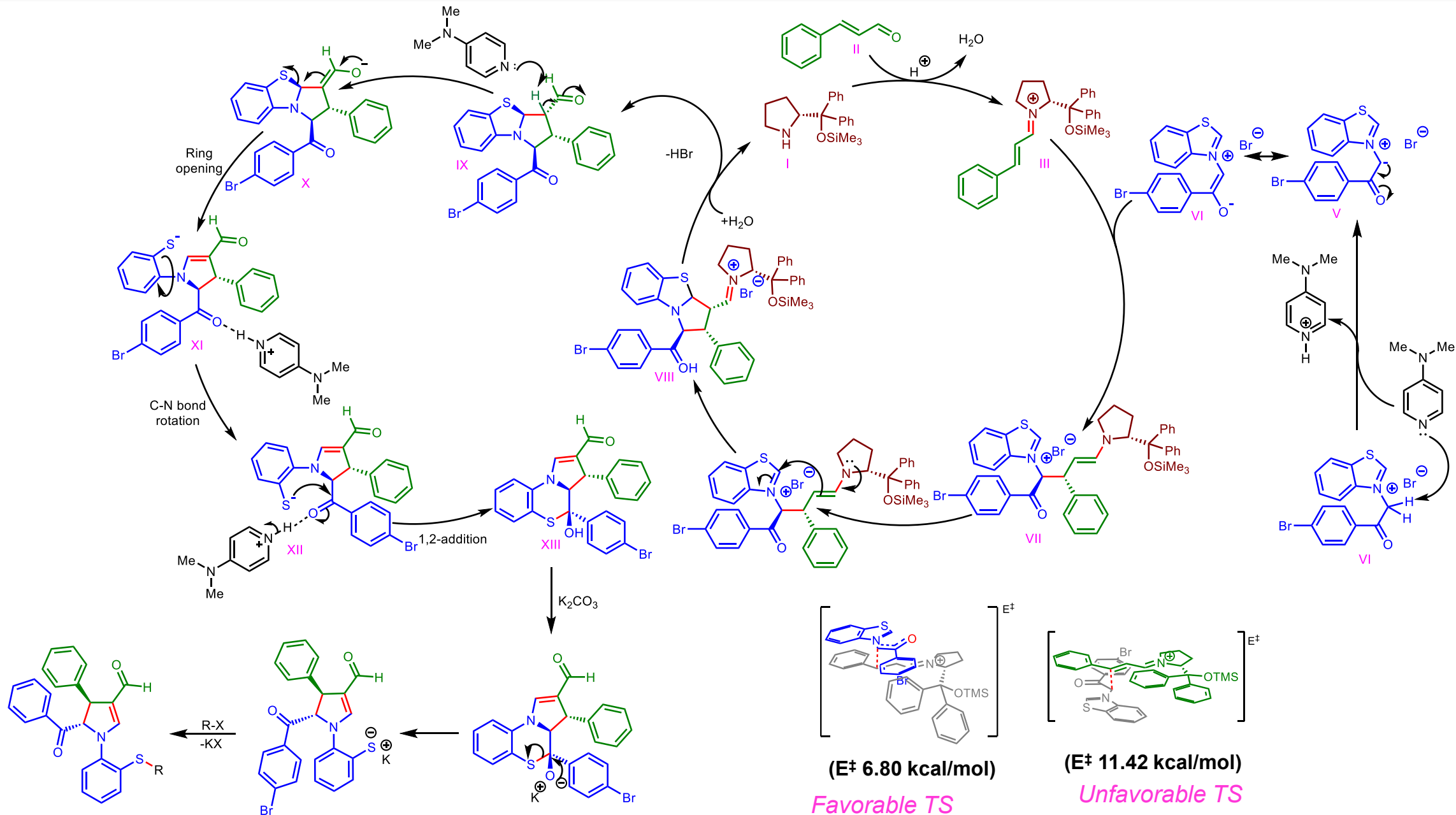
# Computational study



Rotational barrier for C-N bond rotation of intermediate X to form XI computed at M06-2X/6-31G(d,p) level of theory.

Figure 1. Complete reaction profile diagram of the reaction of cinnamaldehyde 1a and benzothiazolium salt 2f with (R)-diphenylprolinol trimethylsilyl ether L3, computed at M06-2X/6-31G(d,p) level of theory.

# Plausible mechanism



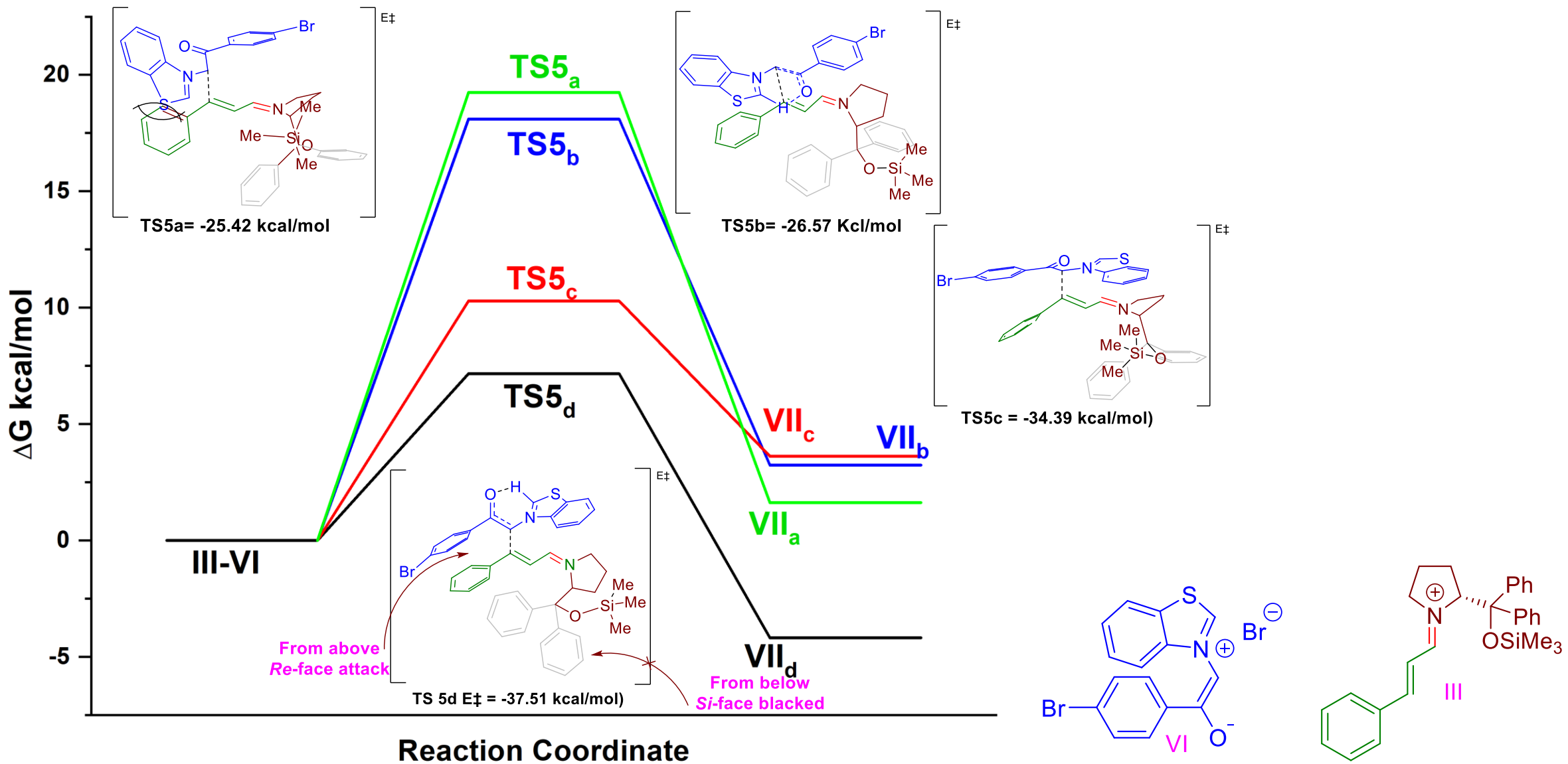
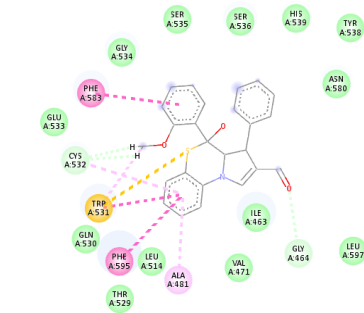
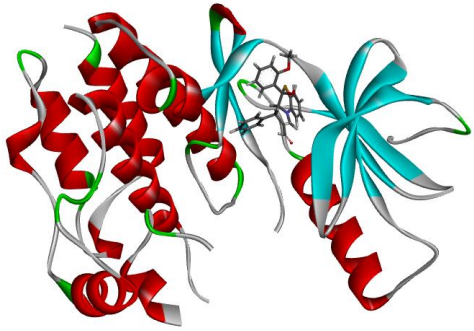


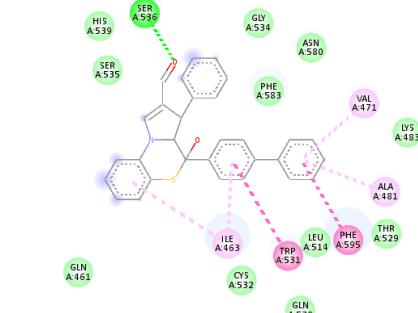
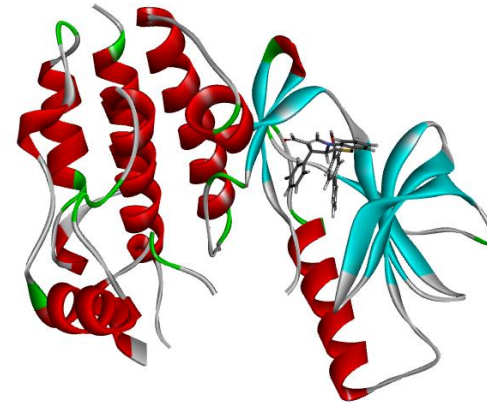
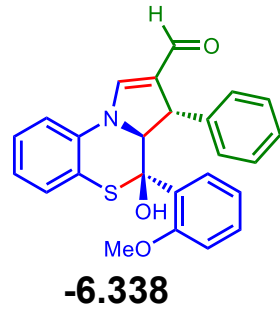
Figure 2. Transition state for C-C bond formation the intermediates **III** and **VI** to combine to produce intermediate **VII** computed at M06-2X/6-31G(d,p) level of theory

# Molecular docking study



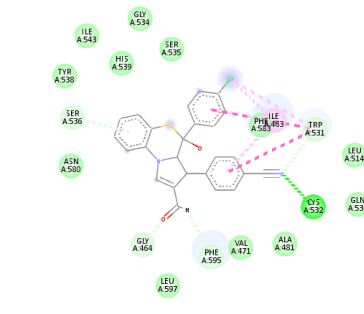
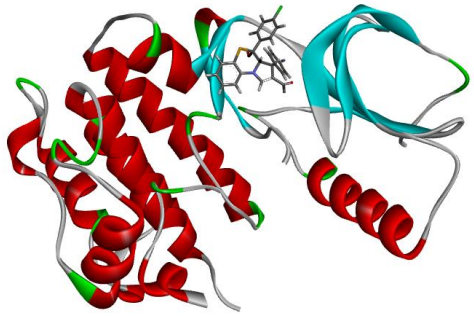
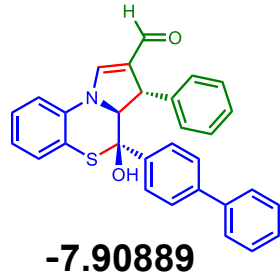
**Interactions**

- van der Waals
- Carbon Hydrogen Bond
- Pi-Sulfur
- Pi-Pi Stacked
- Pi-Pi T-shaped
- Pi-Alkyl



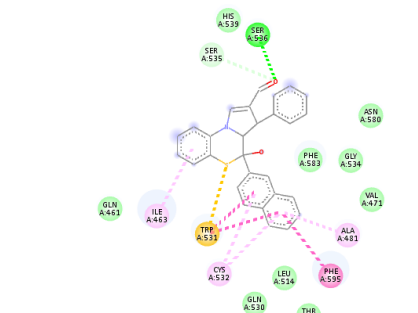
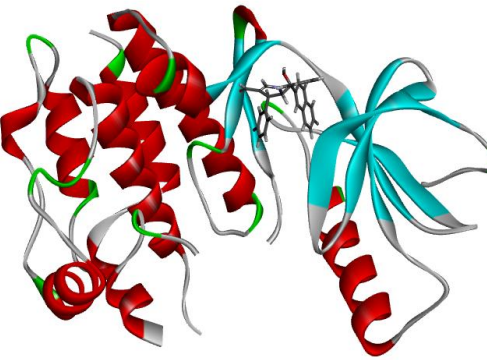
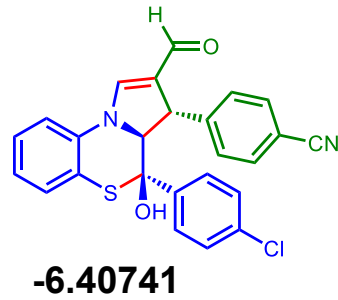
**Interactions**

- van der Waals
- Conventional Hydrogen Bond
- Carbon Hydrogen Bond
- Unfavorable Acceptor-Acceptor
- Pi-Pi Stacked
- Pi-Pi T-shaped
- Pi-Alkyl



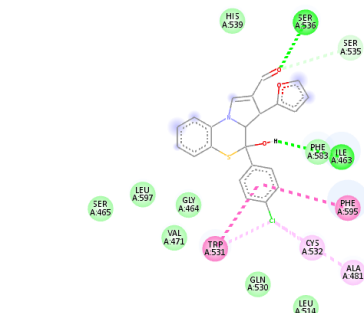
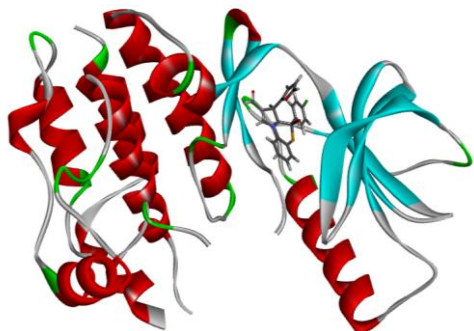
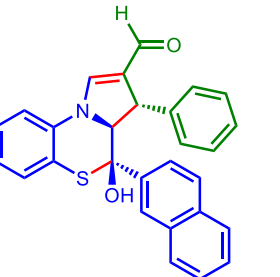
**Interactions**

- van der Waals
- Conventional Hydrogen Bond
- Carbon Hydrogen Bond
- Pi-Donor Hydrogen Bond
- Pi-Pi Stacked
- Pi-Pi T-shaped
- Alkyl
- Pi-Alkyl



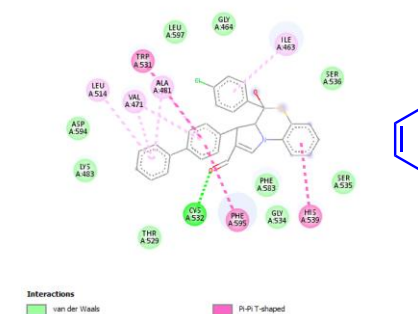
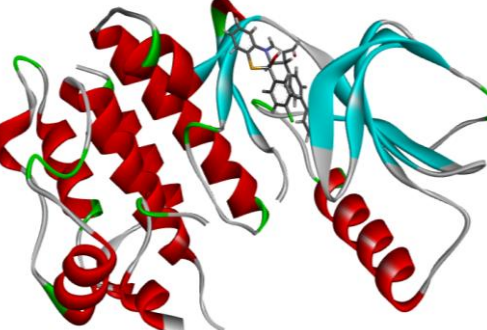
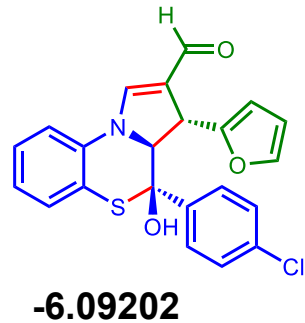
**Interactions**

- van der Waals
- Conventional Hydrogen Bond
- Carbon Hydrogen Bond
- Pi-Sulfur
- Pi-Pi Stacked
- Pi-Pi T-shaped
- Pi-Alkyl



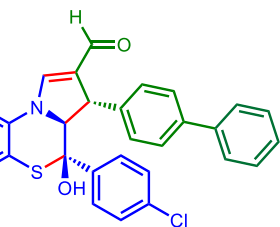
**Interactions**

- van der Waals
- Conventional Hydrogen Bond
- Carbon Hydrogen Bond
- Pi-Pi Stacked
- Pi-Pi T-shaped
- Alkyl
- Pi-Alkyl

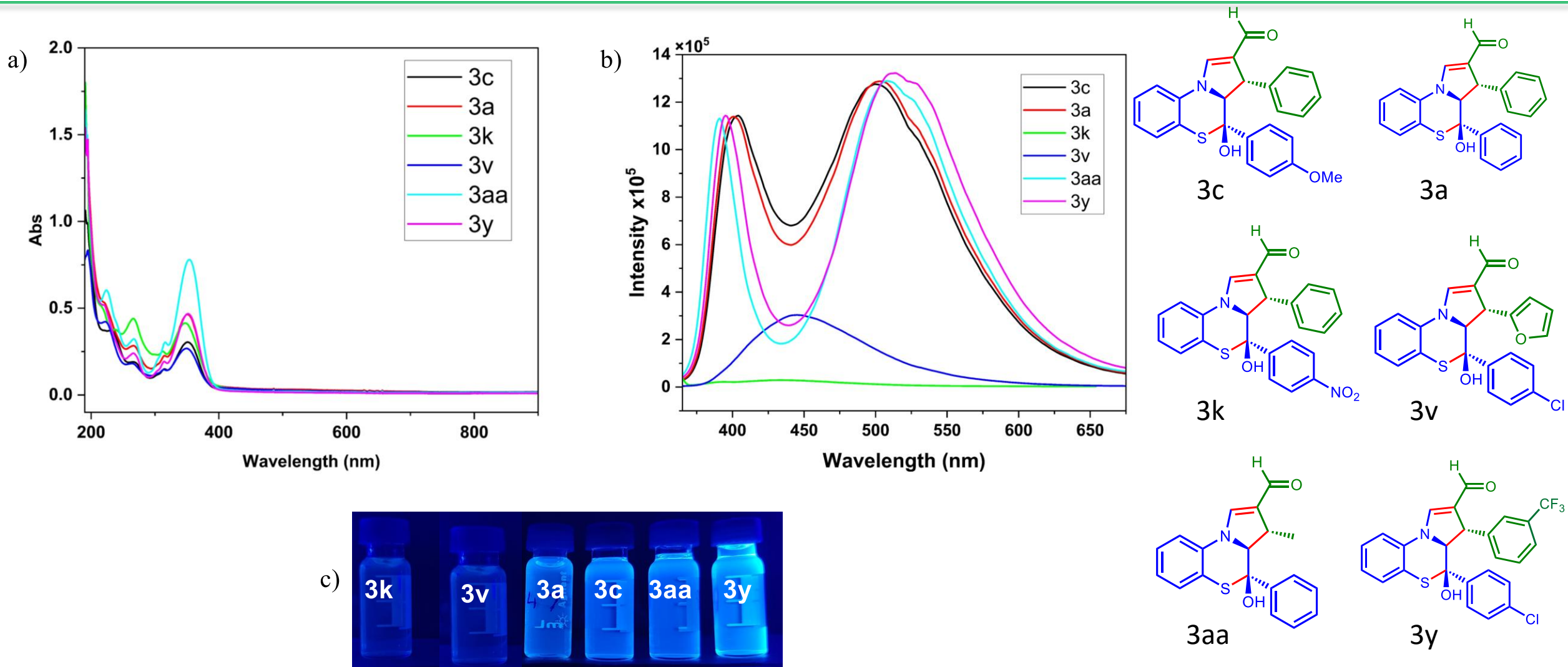


**Interactions**

- van der Waals
- Conventional Hydrogen Bond
- Pi-Pi Stacked
- Pi-Pi T-shaped
- Pi-Alkyl

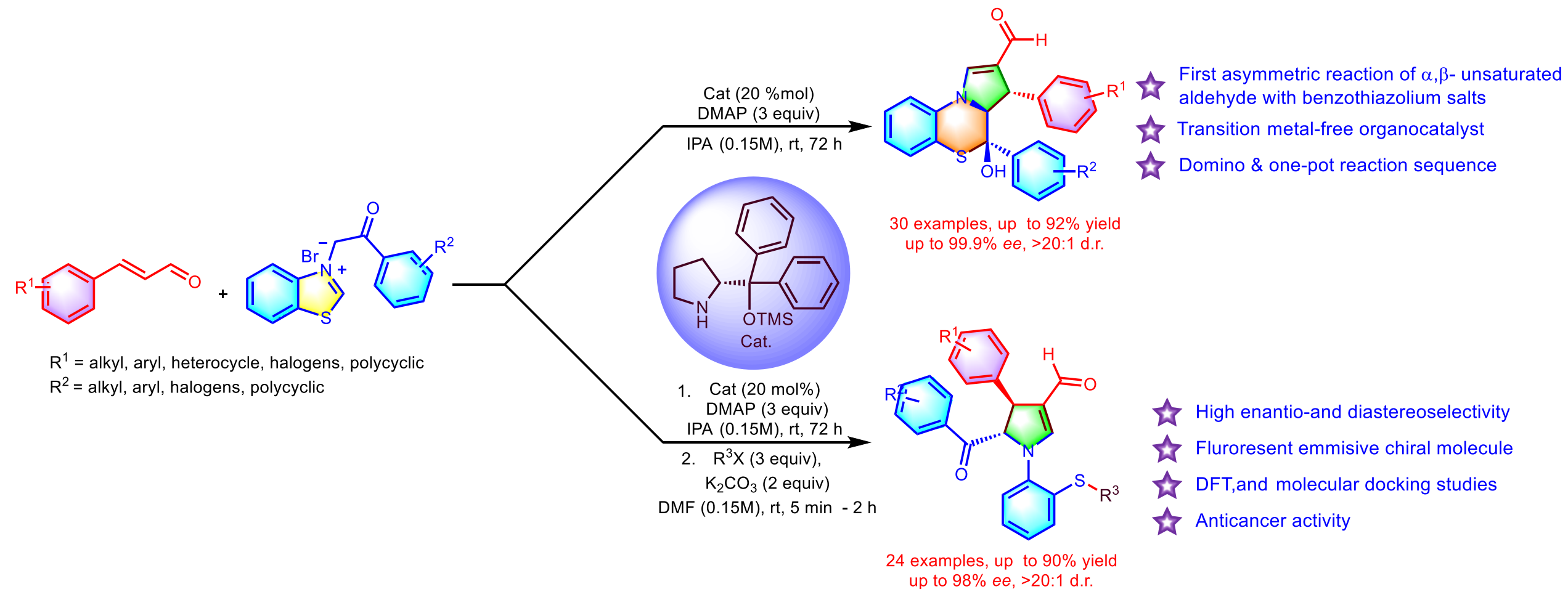


# Fluorescent study



**Figure .1** (a) Absorption spectra in CH<sub>3</sub>CN (50 μm) (b) Fluorescence emission spectra in CH<sub>3</sub>CN (50 μm) (λ<sub>ex</sub> = 352 nm). (c) Fluorescence image in CH<sub>3</sub>CN (λ<sub>ex</sub> = 365 nm).

# Conclusion



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*Thank You*